A feasibility study of using biodegradable magnesium alloy in glaucoma drainage device

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Abstract

● Technological advances in glaucoma have challenged the traditional treatment paradigm. Historically incisional surgery has been used in cases of advanced disease and/or uncontrolled intraocular pressures resistant to medical or laser interventions. More recently, perhaps due to advancements in imaging, surgery has been suggested to be beneficial earlier in the treatment paradigm. Despite these trends, surgical manipulation of the tissues and unpredictability of wound healing continue to result in surgical failure. Magnesium is an essential element for human body and plays a critically important role in maintaining the functional and structural integrity of several tissues, including the eye. Due to several of its advantageous properties such as non-toxicity, biodegradability, and high biological compatibility, magnesium alloy has attracted great attention as a novel biomaterial. Biodegradable cardiovascular stents made of magnesium alloy have already been introduced into clinical practice. The purpose of this review is to determine if bioabsorbable magnesium alloys can be utilized as a promising candidate for the development of a new generation of glaucoma surgical assistive devices.

● KEYWORDS: glaucoma; glaucoma drainage implant; magnesium alloy; anti-scarring

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INTRODUCTION

Magnesium is an essential trace element for human life, and also is one of the most important regulatory cations involved in several biological processes. Magnesium is the second most common cation in the intracellular fluid. Mg2+ plays a crucial role in regulating vascular functions and energy metabolism as well as maintaining water and electrolyte balance[1,2]. The level of Mg2+ in the serum ranges from 0.8 to 1.2 mmol/L, with homeostasis being maintained by the kidney and intestine. Hypomagnesemia (serum Mg2+<0.8 mmol/L) has been suggested to be associated with several disorders, such as vascular spasm and arrhythmia. Magnesium deficiency may also lead to cardiovascular, respiratory, and digestive illnesses, as well as abortion, fetal abnormalities and other obstetric diseases[3]. A large number of studies have shown that dietary intake of magnesium can prevent osteoporosis and femoral neck fracture[4-7], and help to manage diabetes and coronary artery disease[8-9]. Serum Mg2+ levels exceeding 1.2 mmol/L may cause muscular paralysis, respiratory distress and hypotension[10]; however hypermagnesemia is rare due to the efficient excretion of the element in the urin[11].

Glaucoma is the second most common cause of blindness and the leading cause of irreversible worldwide blindness[12]. Surgery is an important treatment option for glaucoma. Incisional surgery, with and without glaucoma drainage devices, is a major part of the treatment paradigm when medical options cannot be tolerated and/or fail to reach a pressure that halts progression. Long-term complications are often related to variability in the wound healing process. Devices used adjunctively in surgical glaucoma management represent a foreign body under the conjunctiva, causing fibroblast proliferation. This may be due to the persistent effect of inflammatory mediators and cytokines at the level of the conjunctival-Tenon-episcleral interface resulting in fibrosis and obstruction of the fistula[13-14]. There continues to be significant interest in adjunctive approaches to minimize scarring in the subconjunctival space, reducing inflammation, and by slowing or halting the excessive fibrotic healing process.
Bleb scarring is a major contributor to increased intraocular pressure (IOP) and surgical failure. In order to improve the success rate of glaucoma surgery, various intraoperative anti-metabolites have been used to inhibit fibroblast proliferation, such as 5-fluorouracil (5-FU) and mitomycin C (MMC). However, these antimitotic agents interfere indiscriminately with cellular proliferation, and can contribute to a number of postoperative adverse events, including bleb leaks, hypotony, choroidal detachment, endophthalmitis, and keratitis\(^{[15-16]}\). There continues to be a significant need for a safe, non-toxic, and effective approach to reduce postoperative scarring and adhesions after glaucoma surgery. The relatively recent introduction of bleb-forming micro-invasive glaucoma surgery has heightened this interest. 

Magnesium alloy is a new kind of biodegradable material which possesses many advantageous properties such as high strength, light weight, and high biological compatibility. Additionally, magnesium alloy has been shown to be able to reduce irritation and inflammatory reactions in the body\(^{[17-20]}\). We were thus motivated to determine the feasibility of developing a magnesium-based bio-absorbable device for modulation of wound healing associated with glaucoma surgery. Moreover, the possibility that the slow release of Mg\(^{2+}\) by the device could have a local neuroprotective role is an intriguing concept worthy of future investigation.

**APPLICATION OF MAGNESIUM ALLOY IN MODERN MEDICINE**

The application of magnesium alloy as a bio-absorbable material in clinical practice can be traced back to the early 1900s. The first use of magnesium was reported by Lambotte\(^{[21]}\) in 1928, who utilized a plate of pure magnesium with gold-plated steel nails to treat a fracture involving the bones of the lower leg. But the attempt failed as the pure magnesium plate corroded too rapidly *in vivo*, disintegrating only eight days after surgery and producing a large amount of gas under the skin. In 1938, McBride\(^{[22]}\) developed magnesium alloy fixtures which successfully treated 20 cases of fracture without any significant adverse effects, and the fixtures were completely absorbed three months after surgery. In 1944, Troitskii and Tsirin\(^{[23]}\) applied cadmium magnesium alloy as an internal fixation device to secure the bones of 34 consecutive fracture patients. They observed that the fixture could maintain the mechanical integrity up to two months, and it was completely absorbed after 10-12mo. Nine cases were unsuccessful, and these failures were attributed to infection. A hard callous was found around the fracture site; no increase in serum levels of magnesium and no obvious inflammatory reactions to the implant were observed in all of the 34 patients. Znamenski\(^{[24]}\) reported similar results in 1945, where magnesium alloy containing 10% aluminum (Al) was used to treat two patients with gunshot wounds. The magnesium implant and nails were completely absorbed after 4-6wk. These early reports demonstrated that magnesium alloy was a non-toxic biomaterial, and it had the ability to promote bone healing. However, due to its characteristics of hydrogen emission and low corrosion resistance in the electrolytic and aqueous environments of the physiological system, biomaterial research on magnesium alloy was suspended. Thereafter, stainless steel materials were widely applied in bone internal fixation devices. Until recent years, with the use of advanced techniques, more complicated alloy compositions have been introduced. Corrosion protection technologies have effectively reduced the production of hydrogen gas. Interest in the medical use of magnesium alloys has once again increased due to advancements in coatings that alter the corrosive tendency\(^{[24]}\). Numerous *in vitro* and *in vivo* studies have focused on the use of magnesium alloys in internal fixation devices for fracture repairs. The corrosion resistance of different types of coated magnesium alloys has been studied in cytological and animal experiments to determine the reduction in hydrogen evolution and tissue compatibility\(^{[25-26]}\). *In vitro* experiments confirmed that there were minimal cytotoxicity and cell apoptosis when fibroblasts and osteoblasts were exposed to various coated magnesium alloys\(^{[26]}\). *In vivo* experiments found that Mg\(^{2+}\) could enhance bone formation during the degradation process of the implant, and no inflammation was observed\(^{[27-29]}\). Witte et al\(^{[29-30]}\) successfully implanted magnesium alloy stents into the femur of rabbits, and the magnesium implants substantially degraded after three months. Their study also showed that the degradation process of the magnesium alloy stents could promote trabecular bone formation and resorption, without any significant harm to their neighboring tissues. They concluded that even fast-degrading magnesium alloy stents could show favourable biocompatibility thus establishing a more convincing potential role, in musculoskeletal surgery. In addition, Witte et al\(^{[31]}\) also carried out an *in vitro* experiment to investigate the properties of a metallic matrix composite made of magnesium alloy AZ91 as a matrix with hydroxyapatite (HA) particles as reinforcements. The results revealed that the HA particles could stabilize the corrosion rate of the magnesium alloy, and this biodegradable MMC-HA was a cytotcompatible biomaterial with adjustable mechanical and corrosive properties. Okazaki\(^{[32]}\) developed a novel material containing magnesium, calcium, and phosphate for use as oral implants. They found that magnesium increased the metabolic rate of osteoblasts. Zreiqat et al\(^{[33]}\) found that the protein levels were significantly higher in human bone-derived cells cultured on (Mg)-Al\(_2\)O\(_3\) (alumina doped with magnesium ions) compared with those grown on Al\(_2\)O\(_3\) alone. Hunt and Heggarty\(^{[34]}\) reported that the magnesium-coated Ti-6Al-4V implant could activate bone cell signal transduction and hence improve protein synthesis and accelerate bone formation. The mechanical and degradation
properties of magnesium have been used to develop novel cardiovascular stents, so as to maintain the endothelial function of coronary arteries and to reduce the risk of coronary ischemia and occlusion\[35-38\]. Based on a large number of clinical trials\[39-40\], magnesium coronary stents have been introduced into clinical practice\[41\].

**THE IMPORTANT ROLES OF Mg\(^{2+}\) IN THE EYE**

Mg\(^{2+}\) is important for maintaining the structural and functional integrity of several vital ocular tissues such as the cornea; lens and retina. The concentration of magnesium in aqueous humor is \(2.97\pm0.75 \text{ mg/100 mL}\). The magnesium levels in the lens are far higher than those in the anterior chamber and vitreous body. The concentration in the lens periphery is four times greater than the axial regions\[62\]. Mg\(^{2+}\) plays an extremely important role in maintaining retinal function. The reason for this is because magnesium acts as a co-factor involved in the catalytic function of more than 350 enzymes in the body and regulates neuro-excitability. Membrane associated ATPase functions that are crucial in regulating the intracellular ionic environment, are also magnesium-dependent. As a result, a reduced level of Mg\(^{2+}\) may affect the functions of Na\(^{+}\)/K\(^{+}\) ATPase and calcium dependent ATPase, causing an increase in the intracellular concentrations of calcium and sodium as well as a decrease in the potassium\[43\] concentration. Such ionic imbalances induced by magnesium deficiency may contribute to the pathogenesis of many eye disorders, such as cataract, corneal, conjunctival, choroidal and retinal diseases\[44-45\].

**Maintainence of Corneal Structure and Function**

Mg\(^{2+}\) is one of the most important cations in the cornea as it is involved in the metabolism and maintenance of corneal transparency\[46\]. As early as 1920, Kirkpatrick\[47\] reported the use of magnesium sulfate in the treatment of keratitis, conjunctivitis, and corneal ulceration. In 1985, Bachman and Wilson\[48\] performed an animal experiment and found that the epithelial surface of the excised rabbit cornea was maintained best with a buffered solution containing Mg\(^{2+}\), K\(^{+}\), and Ca\(^{2+}\). Hogan et al\[49\] reported that Mg\(^{2+}\) loss in the corneal stroma was associated with corneal edema. In 2001, Gong et al\[50\] investigated the effect of magnesium deficiency on the cornea in rats that were fed a low magnesium diet for 3wk. Their findings revealed that magnesium deficiency affected the structural and functional integrity of the cornea.

Keratoconus is defined as a progressive eye disease which causes thinning and fragmentation of membranes, degenerated cells and collagen fibers, swelling of the mitochondria, and biochemical abnormalities in protein synthesis and expansion of central area of the cornea. Thalassellis\[51\] reported that hypomagnesemia was commonly seen in the serum samples of keratoconus patients, and magnesium deficiency could pathologically affect the integrity of the cornea. Prior studies also showed that magnesium deficiency was associated with a reduced number of microvilli and their irregular arrangement. Vesicular degeneration and swelling of the mitochondria were observed in the cytoplasm of epithelial cells, resulting in abnormal apoptosis of these cells\[52-53\]. Magnesium deficiency was also associated with rupture and fragmentation of the Bowman’s layer, which may be an early change leading to keratoconus\[54-55\].

**Supporting Lens Metabolism and Preventing Cataract**

Cataract is the most common cause of blindness worldwide. It is characterized by progressive lenticular opacities. It is known that cataractous lenses have an abnormal intracellular ionic environment with lower concentrations of potassium and magnesium and higher concentrations of sodium and calcium relative to the cytosol of most cells. The lens membrane has increased permeability in the presence of a cataract\[60\]. Studies have explored the relationship between magnesium deficiency and cataract. These studies have shown that the alterations in lenticular redox status and ionic imbalances form the basis of the relationship between magnesium deficiency with cataract.

It is believed that Mg\(^{2+}\) plays an important role not only in maintaining a low lenticular Ca\(^{2+}\) and Na\(^{+}\) concentration but also in preserving the lens redox status, which has been shown to reduce lenticular oxidative stress\[57-58\]. Nagai et al\[59\] reported that the incidence of cataract was significantly lower in Shumiya rats fed 200 mg/L magnesium compared to controls. It was noted that the intracellular Ca\(^{2+}\) level of the lenses was also significantly lower. Based on these findings, they concluded that an appropriate supplementation of magnesium could delay cataract genesis, probably by inhibiting the increase in Ca\(^{2+}\) levels in the lens\[60\]. After a thorough literature review, Agarwal et al\[61\] concluded that magnesium supplementation might be of therapeutic value in preventing the onset and progression of cataract.

Oxidative stress is another important contributor to the pathogenesis of cataract. It has been shown in many studies that nitric oxide (NO) could be a risk factor in cataract formation\[62\]. NO production via inducible nitric oxide synthase (iNOS) causes an oxidative stress response in the lens\[63\]. Mg\(^{2+}\) has been shown to prevent the increase of NO in the lens which provides protection by inhibiting the nitrosylation of gap junctional proteins and maintaining membrane permeability\[64\]. It is also known to block iNOS expression and reduce the oxidative stress response. A cytoplological experiment found that the expression of iNOS was 6 times higher in the lens epithelial cells cultured in the magnesium-deficient medium compared to those cultured in the magnesium-supplemented medium. In addition, Mg\(^{2+}\) was shown to regulate Ca\(^{2+}\) ATPase thus modulating the concentration of Ca\(^{2+}\) in the lens.

Both magnesium deficiency and excessive NO production have been shown to decrease ATP levels\[60\]. The reduction in adenosine triphosphate ATP levels affects the membrane-
associated Ca$^{2+}$ ATPase and Na$^+$/K$^+$ ATPase, causing ionic imbalance in the lens. Interestingly, the decreased ATP levels in turn may have inhibitory effects on the expression of iNOS[80]. Therefore, it has been speculated that magnesium deficiency may cause an acceleration of the progression of lens opacification (Figure 1).

Glaucomatous Neuroprotection Elevated IOP is not the sole risk factor for chronic glaucomatous neuropathy. In fact, the search for putative non-IOP dependent continues to be of significant interest. Vasomotor dysfunction has also been suggested to contribute to optic neuropathy, by mediating abnormal hemodynamics and oxidative stress.[66]. In an epidemiological study, Bonomi et al[67] reported that reduced diastolic perfusion pressure could be an important risk factor for primary open-angle glaucoma. Leske et al[68] reported an association between cardiovascular disease and glaucoma which suggested a vascular role in glaucomatous progression.

It has been well-established that Mg$^{2+}$ can function as a physiological Ca$^{2+}$ channel blocker which, in turn, prevents ischemia and provides protection to the optic nerve. A large number of studies concerning ocular blood flow and oxidative stress response have confirmed that Mg$^{2+}$ may regulate the current strength and in activation process of Ca$^{2+}$ channels.[69-70]. Even small changes in extracellular Mg$^{2+}$ levels may have significant effects on vascular tone. Mg$^{2+}$ has been shown to have a direct vasodilatory effect[71-73] while magnesium deficiency has been shown to increase intracellular Ca$^{2+}$ levels leading to vasoconstriction[74-75]. Moreover, as Mg$^{2+}$ regulates Na$^+$/K$^+$ ATPase, magnesium deficiency is associated with reduced activity of Na$^+$/K$^+$ ATPase and increased intracellular Na$^+$ and Cl$^-$ levels causing cellular swelling and apoptosis of retinal ganglion cells (RGCs).[76].

Other studies have demonstrated an association between plasma endothelin-1 (ET-1) levels and normal tension glaucoma (NTG). Patients with NTG in the initial stage of visual field loss demonstrated higher plasma ET-1 levels than those with moderate visual field damage.[77]. An increase in extracellular Mg$^{2+}$ levels inhibits ET-1, which may induce constriction and vasospasm of the ciliary arteries, with reduced blood supply to the optic nerve. Furthermore, ET-1 may affect the functions of axons and astrocytes, and accelerate the apoptosis of RGCs.[73].

It has been suggested that the favorable effects of magnesium on the visual field may be attributed to ET-1-inhibition, suppressing vasoconstriction, improving the ocular blood flow, and preventing glaucomatous neuropathy.[44].

The ion channel of the N-methyl-D-aspartate (NMDA) receptor is calcium dependent and subject to voltage-dependent regulation by Mg$^{2+}$. Mg$^{2+}$ can regulate the glutamate-gated ion channel which, in turn, has been shown to prevent excitotoxicity and cell apoptosis.[70-82]. In the presence of a magnesium deficiency, the toxic effects of glutamate on RGCs are mediated by the

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Magnesium deficiency

- Dysfunction of Ca$^{2+}$ ATPase
- Reduced ATP production
- Increased NO expression
- Intracellular accumulation of Ca$^{2+}$
- Osmotic stress
- Intracellular accumulation of Na$^+$
- Oxidative stress
- Cataract

Figure 1 Magnesium deficiency contributes to the pathogenesis of cataract.

over stimulation of NMDA receptors, leading to glutamate excitotoxicity and loss of RGCs in glaucoma patients. Lambuk et al[83] reported that intravitreal Mg$^{2+}$ could prevent retinal and optic nerve damage induced by NMDA. Moreover, intracellular accumulation of Ca$^{2+}$ associated with magnesium deficiency has been associated with the production of free radicals.[84-85].

Magnesium deficiency directly increases the expression of iNOS. Lower levels of Mg$^{2+}$ have been associated with vasospasm and retinal ischemia which also enhance the expression of iNOS[86], and produce large amounts of free radicals[50]. Numerous studies have shown that Mg$^{2+}$ plays a neuroprotective role in inhibiting the elevated iNOS activity of neurons in retinal ischemia.[87-90].

In a study by Gaspar et al[90], 10 glaucoma patients including 6 with primary open angle glaucoma (POAG) and 4 with NTG were administered magnesium 121.5 mg, twice a day for one month. Results showed that magnesium significantly increased the ocular blood flow and improved the peripheral circulation, exerting a beneficial effect on the visual field in glaucoma patients with vasospasm. Aydin et al[91] also reported that in 15 NTG patients who received 300 mg oral magnesium citrate for a month, an improvement in the visual field was observed, but the ocular blood flow remained unchanged. They speculated that mechanisms other than increased ocular blood flow may be responsible for the improvement in the visual field when given oral magnesium therapy. Based on these findings, it was suggested that Mg$^{2+}$ may play an important role in optic neuroprotection in patients with glaucoma (Figure 2).

The History and Prospects of Using Magnesium Materials in Glaucoma Drainage Surgery

Due to the unique properties of magnesium and its satisfactory bio-compatibility, the use of the pure materials was considered
as an implant for the treatment of glaucoma in the middle of the 20th century. In 1940, Troncoso[92] utilized pure magnesium implants to increase the aqueous outflow in the treatment of glaucoma. Five years later, Hoff[93] used this technique to treat a patient with neovascular glaucoma who refused enucleation. Twelve hours after operation, the patient’s whole anterior chamber was filled with gas and the IOP was significantly increased, resulting in severe pain. The pressure of the gas was relieved by venting with a thin hypodermic needle. The IOP normalized on the sixth postoperative day, but huge bubbles were still found in the anterior chamber and the conjunctiva, which disappeared on the tenth day. Interestingly, before the operation the cornea was opaque at an IOP of 54 mm Hg, but it became absolutely clear when the anterior chamber was full of gas, despite the elevated IOP which varied from 49 to 55. The postoperative IOP was controlled at 1-4mo after surgery, but elevated to 45 mm Hg again at the fifth month. Gonioscopy revealed complete circumferential iridocorneal synechiae. Coarse pigment granules were observed in the dependent part of the angle. Eventually, the eye was removed. Pathological examination revealed a closed filter channel and massive fibroblast proliferation around the incision.

These experiences indicated that pure magnesium had significant adverse effects due to the excessive corrosion when in contact with body fluids with hydrogen gas emission causing treatment failure. Advancements in the area of coating technology, allowed the synthesis of novel coated magnesium alloys with optimized compositions (containing Al, zinc, manganese, calcium, praseodymium, or neodymium) allowing control over the corrosive properties. Coated magnesium alloys have shown significant promise as effective biodegradable materials.

During the last 15y, biodegradable metallic stents have been developed and investigated as alternatives for the currently-used permanent cardiovascular stents[94-96]. Traditional cardiovascular stents have significant drawbacks including irritation and damage to the vascular endothelium, leading to stenosis and blockage[97-100]. In order to solve these problems, absorbable magnesium materials have shown success. In vitro studies have demonstrated that magnesium cardiovascular stents can maintain the integrity and function of endothelial cells, as well as reduce inflammation and fibroblast proliferation. In these studies, stents could be fully absorbed within 2-4mo with few complications observed during a two years of follow-up period[38,40]. Subsequently, a large number of clinical trials were conducted[38-39]. In 2007, a prospective multi-center trial in patients with coronary heart disease was published in the Lancet, which showed that biodegradable magnesium stents achieved an immediate angiographic result similar to the result of traditional metal stents and was safely degraded after 4mo[101].

The introduction of magnesium cardiovascular stents has led to interest in how their use can be expanded into other areas of the body. In particular, their potential utility as an adjunctive device in glaucoma surgery is of high interest. The excellent biocompatibility and bio-degradability, have the potential to allow magnesium alloy devices to effectively decrease postoperative irritation, reduce scar formation, improve the success rate of glaucoma surgery, and minimize the long-term effects of permanent implants which tend to produce local complications related to tissue compression. The liberation of Mg2+ cations during the natural degradation process has many potential beneficial effects on the cornea, lens, retina, choroid, and optic nerve.
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In conclusion, bio-absorbable coated magnesium alloys may be a very promising candidate for the development of a new generation of glaucoma drainage device.

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